

In-class problem linked to lecture pages 71-84:

Consider 2 interacting systems for which $\Omega_1 = 2 \times 10^{10^{24}}$ and $\Omega_2 = 4 \times 10^{10^{24}}$.

- a) How many states are accessible to the combined system?
- b) What are the states' individual entropies, in terms of k ?
- c) What is the entropy of the combined system, in terms of k ?

In-class problem linked to pages 71-82

$$\Omega_1 = 2 \times 10^{10^{24}}$$

$$\Omega_2 = 4 \times 10^{10^{24}}$$

$$\begin{aligned} \text{a) } \Omega_0 &= \Omega_1 \cdot \Omega_2 = 2 \times 4 \times 10^{10^{24}} \\ &= 8 \times (10^{10^{24}})^2 \\ &= 8 \times (10^2)^{10^{24}} \\ &= 8 \times 10^{2 \times 10^{24}} \end{aligned}$$

$$\text{b) } S_1 = k \ln [2 \times 10^{10^{24}}]$$

$$\frac{S_1}{k} = \ln_e [2 \times 10^{10^{24}}] = \log [2 \times 10^{10^{24}}] \cdot \ln(10)$$

$$\begin{aligned} &= \log 2 + \log 10^{10^{24}} \ln 10 \\ &= \text{neglect } (10^{24}) \cdot 2.3 \\ &= 2.3 \times 10^{24} \end{aligned}$$

$$\frac{S_2}{k} = \ln_e [4 \times 10^{10^{24}}] = \log [4 \times 10^{10^{24}}] \ln(10)$$

$$\begin{aligned} &= \log 4 + (10^{24}) \cdot 2.3 \\ &= 10^{23} \cdot 2.3 \\ &= 2.3 \times 10^{24} \end{aligned}$$

$$\text{c) } \frac{S_{\text{TOT}}}{k} = \frac{S_1 + S_2}{k} = 4.6 \times 10^{24}$$