

In-class problem linked to lecture pages 176 – 190

Suppose that you want to predict how energy varies with temperature and pressure during a non-diffusive process. Show that:

$$\left. \frac{\partial E}{\partial p} \right|_T = V(p\kappa - T\beta)$$

and

$$\left. \frac{\partial E}{\partial T} \right|_p = C_P - pV\beta$$

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$$\text{Generally } dE = TdS - PdV + \mu dN$$

Here $dN = 0$

$$\text{Let } dS = \left. \frac{dT dS}{dT} \right|_P + \left. \frac{dp dS}{dp} \right|_T$$

$$\text{Let } dV = \left. \frac{dT dV}{dT} \right|_P + \left. \frac{dp dV}{dp} \right|_T$$

Then

$$dE = T \left. \frac{dS}{dT} \right|_P dT + T \left. \frac{dS}{dp} \right|_T dp - P \left. \frac{dV}{dT} \right|_P dT - P \left. \frac{dV}{dp} \right|_T dp$$

Eg1.

$$= \left[T \left. \frac{dS}{dT} \right|_P - P \left. \frac{dV}{dT} \right|_P \right] dT + \left[T \left. \frac{dS}{dp} \right|_T - P \left. \frac{dV}{dp} \right|_T \right] dp$$

Note Maxwell Relation M10 (Stove p. 217):

$$\left. \frac{dS}{dp} \right|_{T,N} = \left. \frac{dV}{dT} \right|_{P,N} = V\beta \quad (\beta = \text{coef of volume exp, Stove p. 169})$$

$$\text{If } dQ = TdS, \text{ then } \left. \frac{dS}{dT} \right|_P = \frac{1}{T} \left. \frac{dQ}{dT} \right|_P = \frac{1}{T} C_P$$

$$\text{Isothermal compressibility } K = \left. \frac{1}{V} \frac{dV}{dp} \right|_T$$

$$\text{So } \left. \frac{dV}{dp} \right|_T = -KV$$

Plug into Eq 1 to get:

$$dE = \left[T \cdot \frac{1}{T} C_p - P V \beta \right] dT + \left[T(-V\beta) - P(-KV) \right] dp$$

$$= [C_p - P V \beta] dT + [P K V - T \beta V] dp \quad \text{"physics eq"}$$

Compare this to the "math eq":

$$dE = \frac{dE}{dT} \Big|_p dT + \frac{dE}{dp} \Big|_T dp \quad \text{to conclude:}$$

$$\frac{dE}{dT} \Big|_p = C_p - P V \beta$$

$$\frac{dE}{dp} \Big|_T = (P K - T \beta) V$$